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# Physico-chemical characterization and antibacterial activity of inclusion complexes of *Hyptis martiusii* Benth essential oil in $\beta$ -cyclodextrin



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## ABSTRACT

Cyclodextrins (CDs) have been used as important pharmaceutical excipients to improve the physicochemical properties of drugs of low solubility as the essential oil of *Hyptis martiusii*. This oil is important therapeutically, but the low solubility and bioavailability compromise your use. Therein, the aim of this study was to obtain and to characterize physico-chemically the samples obtained by physical mixture (PM), paste complexation (PC) and slurry complexation (SC) of the essential oil *Hyptis martiusii* (EOHM) in  $\beta$ -CD, and to compare the antibacterial and modulatory-antibiotic activity of products obtained and oil free. The physicochemical characterization was performed by differential scanning calorimetry (DSC), thermogravimetry/derivative thermogravimetry (TG/DTG), scanning electron microscopy (SEM), X-ray diffraction (XRD) and Karl Fischer titration. Additionally, the antibacterial tests were performed by microdilution technique. Thus, it was observed that the PM method showed low complexing capacity, unlike PC and SC in which it was observed the formation of inclusion complexes. In addition, the second stage of the TG/DTG curves showed that SC was the best method inclusion with mass loss of 6.9% over the PC that was 6.0%. The XRD results corroborate with the results above suggesting the formation of new solid phase and the SEM photomicrographs showed the porous surface of the samples PC and SC. The essential oil alone demonstrated an antibacterial and modulatory effect against the *S. aureus* and the Gram negative strain, respectively. However, the  $\beta$ -CD and the inclusion complex did not demonstrate any biological activity in the performed antibacterial assays.

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## 1. Introduction

The *Hyptis* genus belongs to the Lamiaceae family and is used in the folk medicine as an alternative therapy in the diseases treatment. Some of these species, such as *Hyptis fruticosa*, *Hyptis*

*suaveolens*, *Hyptis pectinata*, are characterized by the presence of essential oils with pharmacological properties, such as antibacterial [1], antiseptic [2], antiulcer [3], antinociceptive [4], anti-inflammatory [5,6], among others.

The species *Hyptis martiusii* Benth, known as “cidreira do campo” or “cidreira brava” is a small shrub, commonly found in the northern, southeastern and northeastern Brazil [7]. In folk medicine, infusion or decoction of *Hyptis martiusii* leaves are used to combat diseases of the gastrointestinal tract while the decoction of the root misused to combat inflammation of ovaries [8]. In studies with *Hyptis martiusii* species were identified a number of

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activities, including cytotoxic and antiproliferative effects in certain tumor cell lines [9,10], insecticidal activity [11], antimicrobial [12] and antiulcerogenic [13], highlighting the pharmacological potential of this specie.

In this context, essential oils can be defined as complex mixtures of volatile organic compounds produced as secondary metabolites of plants. Many of them have demonstrated antibacterial activity, alone or associated with antibiotics, by different mechanisms, diffculting the bacterial adaptation and avoiding the resistance surveillance [14,15]. They are made up of hydrocarbons and oxygenate compounds [16]. In the literature, the *Hyptis martiusii* essential oil (EOHM) is described as comprised of mono- and sesquiterpenes, and has as major component 1,8-cineole (24.3%),  $\delta$ -3-carene (22.5%), bicyclogermacrene (6.3%) and  $\beta$ -caryophyllene (6.2%) [17].

Thus, the incorporation of volatile oils in cyclodextrins has been applied to protect the compounds against temperature, oxidation, evaporation and humidity [18]. The use of complexing molecules can be used to combat the bacterial resistance also, reducing the concentration of the drugs and the necessary doses used in the treatment [19]. Besides, the use of molecules with characteristics of to form inclusion complex as CDs associated with antimicrobial drugs can be enhanced the water solubility and extend the half-life in the system of these drugs [20].

Cyclodextrins (CDs) are oligosaccharides composed of 6, 7, 8 or 9 glucopyranose units ( $\alpha$ -,  $\beta$ -,  $\gamma$ - or  $\delta$ -CD, respectively), with a relatively hydrophilic surface and a hydrophobic central cavity [21,22]. The cyclodextrin inclusion complex formation has been successfully applied to enhance the chemical stability, solubility and bioavailability of poorly soluble compounds [22–24]. The minimum requirement for the complex to be formed is the compatibility of sizes and geometries between the CD cavity and the guest. It has also to consider the hydrophobic character of the guest, thus the polarity is a conditioning factor in the formation of the inclusion complex [25].

The objective of this study was to elucidate the complexation of EOHM in  $\beta$ -CD, by different methods, and to evaluate the antibacterial and modulatory-antibiotic effect of the best formulation obtained in comparison of oil free. The samples were characterized by differential scanning calorimetry (DSC), thermogravimetry/derivative thermogravimetry (TG/DTG), scanning electron microscopy (SEM), X-ray diffraction (XRD) and moisture content determined by Karl Fischer titration method.

## 2. Material and methods

### 2.1. Material

The  $\beta$ -CD (Lot:#041M1759V; purity  $\geq 97\%$ ) was purchased from Sigma-Aldrich (USA) and the EOHM was extracted by hydrodistillation of *Hyptis martiusii* leaves collected in the savannah area of the Chapada do Araripe (Barreiro Grande Farm, Crato-Ce, 7°21'50" S; 39°28'39" W, elevation: 930 m) in May/2012. A voucher of the plant specimen was deposited in the Carirense Dárdano de Andrade Lima Herbarium – HCDAL of the Regional University of Cariri – URCA, under registration number 8394. The plant was identified and classified by Prof. Maria Arlene Pessoa da Silva.

### 2.2. Sample preparation

#### 2.2.1. Extraction of the essential oil

Fresh leaves samples (3.950 g) from their natural habitat were collected, washed in running water, pulverized submitted to hydrodistillation (4 h) using a modified clevenger-type apparatus. The biphasic mixture is formed by the essential oil in the superior phase and the aqueous phase, which is separated through

decantation, dried over anhydrous  $\text{Na}_2\text{SO}_4$ , kept in amber bottle flask and maintained in temperature lower than 4 °C. The yield (0.34%, w/w) of essential oil was calculated based on oil volume produced and fresh leaves mass in kg. The oil of yellowish coloration and characteristic odor was evaluated.

## 3. Complexation with $\beta$ -cyclodextrins ( $\beta$ -CD)

The samples were prepared by techniques of physical mixture (PM), paste complexation (PC) and slurry complexation (SC). Thus, in the PM method, EOHM (154 mg, based on the molecular weight of the major component, 1,8-cineole) and the  $\beta$ -CD (1135 mg) were mechanically mixed by 10 min in the molar ratio of 1:1 in ambient temperature. For to obtain the PC, the EOHM (154 mg) and  $\beta$ -CD (1135 mg) were mixed with the aid of a mortar and pestle (1:1 molar ratio) and adding then 2.0 mL of distilled water with constant manual stirring until forming a paste. Then, the material was kept in a desiccator until dry. Finally, the SC was performed by mechanical mixture of EOHM (154 mg) and  $\beta$ -CD (1135 mg) in a molar ratio of 1:1 and then 20 mL of water was added, and remained under constant magnetic stirring at 150 rpm for 36 h. Subsequently, the material was stored in a desiccator until dry.

### 3.1. Differential scanning calorimetry (DSC)

The EOHM,  $\beta$ -CD, PM and of inclusion complexes samples were subjected to DSC test. The DSC curves were obtained using a DSC-50 cell, from Shimadzu, using a heating rate of 10 °C/min. The DSC curves were obtained between 25 and 500 °C under a dynamic atmosphere of  $\text{N}_2$  (50 mL/min), employing aluminum capsules (Al) containing  $\sim 2$  mg of the samples. The DSC cell was calibrated with indium (melting point 156.6 °C;  $\Delta H_{\text{fus}} = 28.54$  J/g) and zinc (melting point 419.6 °C).

### 3.2. Thermogravimetry/derivative thermogravimetry (TG/DTG)

The EOHM,  $\beta$ -CD, PM and inclusion complexes samples were tested in a TG/DTG apparatus. The TG/DTG curves were obtained using TGA-51 thermobalance, from Shimadzu, using a heating rate of 10 °C/min. The TG/DTG were conducted at 25–900 °C range under a dynamic atmosphere of  $\text{N}_2$  (50 mL/min) using platinum (Pt) capsules containing  $\sim 3$  mg of the samples. The thermogravimetric system was verified using  $\text{CaC}_2\text{O}_4 \cdot \text{H}_2\text{O}$ , reference substance, in accordance with the ASTM standard.

### 3.3. Determination of moisture content by the Karl Fischer method

The moisture content of the PM and the inclusion complexes were determined by Karl Fischer method, model Titrino Plus KF870 (Metrohm), and methanol (Fluka) was used as a titration solution. Analyses were performed in triplicate.

### 3.4. Scanning electron microscopy (SEM)

The  $\beta$ -CD, PM, PC and SC samples were mounted on aluminum stubs subsequently plated with gold beams and viewed in an electron microscope (JEOL JSM-6390-LV model) in acceleration voltage of 12 kV.

### 3.5. X-ray diffraction (XRD)

The X-ray diffraction of the products were obtained in a Siemens D 5000 equipment model with  $\text{CuK}_\alpha$  tubes, in the range of 3–65° ( $2\theta$ ) and 1s at each step.

### 3.6. Microbial strains

The bacteria used in the assays were obtained in the Laboratory of Microbiology and Molecular Biology – LMBM, of Regional University of Cariri – URCA. So, it was used Gram positive (*Staphylococcus aureus* ATCC 25923 and a multidrug resistant *S. aureus* 10) and Gram negative strains (*Escherichia coli* ATCC 25922, *Pseudomonas aeruginosa* ATCC 9027 and a multidrug resistant *E. coli* 06 and *P. aeruginosa* 15). The resistance profile was previously described by Lima et al [26].

### 3.7. Antibacterial and antibiotic assays

For this assay, it was performed the microdilution method using microtiter plates with 96 wells [27]. The assays of antibacterial and modulatory-antibiotic activity were performed as reported by Coutinho et al. [12]. All assays were performed in triplicate and using the follow compounds: EOHM,  $\beta$ -CD and the inclusion complex obtained by SC method with concentrations ranging between 1024 and 1  $\mu$ g/mL. The microtiter plates were incubated by 24 h at 37 °C. The Minimum Inhibitory Concentration (MIC) was determined using 20  $\mu$ L of resazurine 0.01% on each well and observing the change of colour. The MIC is defined as the lower concentration that inhibits the bacterial growth [28].

### 3.8. Statistical analysis

The bacterial results were analysed using the geometric means followed by ANOVA two-way with Bonferroni post hoc test ( $P < 0.05$  is considered significant).

## 4. Results and discussion

### 4.1. Complexation of EOHM in $\beta$ -CD

The DSC curve of EOHM showed an endothermic event at temperature range 28–140 °C corresponding to its volatilization (Fig 1). The  $\beta$ -CD curve showed three endothermic events followed by an exothermic and subsequent decomposition step. The endothermic events occurred in the range 32–121, 205–240 and 285–344 °C. The first endothermic event was related to water loss of  $\beta$ -CD [29] and the second endothermic event was a physical process attributed to the change of crystalline phase [30]. Then there was observed melting followed by degradation of  $\beta$ -CD.

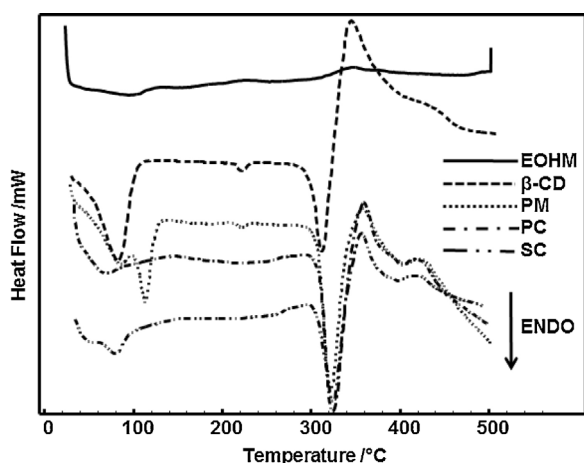


Fig. 1. DSC curves in dynamic  $N_2$  atmosphere of EOHM,  $\beta$ -CD, PM and the complex EOHM: $\beta$ -CD obtained from PC and SC methods.

The DSC curve of PM presented four endothermic events followed by decomposition. The events occurred in the following temperature ranges: 67–97, 99–132, 212–234 and 292–356 °C. The first two events represented a sum of EOHM and  $\beta$ -CD. The third peak corroborated crystalline phase transition of  $\beta$ -CD and finally, the melting of  $\beta$ -CD followed by decomposition.

DSC curves of PC and SC were different from PM and isolated raw materials and showed endothermic peaks followed by decomposition. In the PC events occurred in the following temperature range: 30–148 and 295–358 °C. In the SC curve, the endothermic events occurred at 35–61, 62–134 and 297–356 °C. In curves of these last two methods were observed significant differences mainly related to the reduction of the intensity of the peak corresponding to loss of water and disappearance of the endothermic event related to crystalline phase transition of the pure  $\beta$ -CD.

The formation of inclusion complexes was also studied by TG/DTG curves of EOHM,  $\beta$ -CD, PM, PC and SC samples. Fig. 2 shows the TG/DTG curves of the samples and Table 1 lists mass losses calculated from specific intervals for each material studied in this paper and water percentages calculated by the Karl Fischer method.

The TG results showed that the EOHM volatilized up to 170 °C, with a weight loss of 100% as showed in the DSC technique. In the same temperature  $\beta$ -CD showed mass loss of 12.2% ( $\%_{H_2O} = 13.42$ ) attributed to the molecule dehydration (Step 1). Furthermore, between 170 and 280 °C, there was no significant weight loss, characterizing the crystalline phase transition described in the corresponding DSC curve. Thermal decomposition occurred after 280 °C and through the DTG curve, it was confirmed that the maximum decomposition temperature of  $\beta$ -CD is 343 °C [31]. In PC and SC curves, it was observed that in the range of 170–280 °C (Table 1) the mass losses were 6.0 and 6.9%, respectively. These weight loss percentages were attributed to release of complexed EOHM, since the free oil and  $\beta$ -CD had no significant weight loss at this stage. In addition, the PM curve showed a low percentage of weight loss in the same step, suggesting that this preparation method is not effective in the complexation of EOHM in the  $\beta$ -CD cavity. Furthermore, through the percentage of water values obtained by Karl Fischer titration, as shown in Table 1, it was observed a decrease in the percentage of water in the complex when compared with the  $\beta$ -CD. This decrease can be indicative of the formation of the inclusion complex, since the water molecules of  $\beta$ -CD cavity were replaced by guest molecules [32].

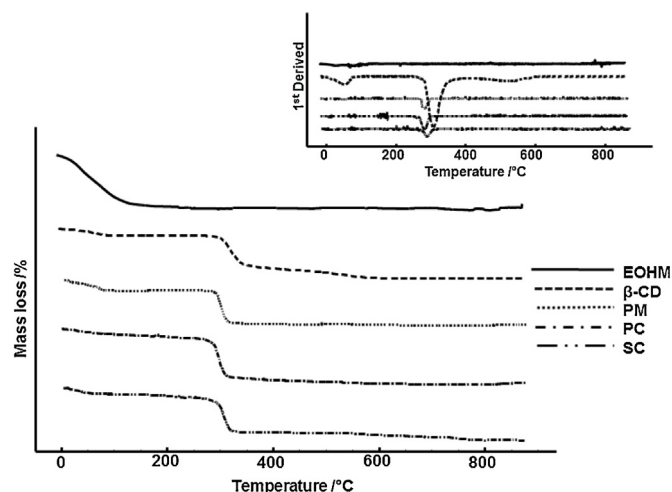


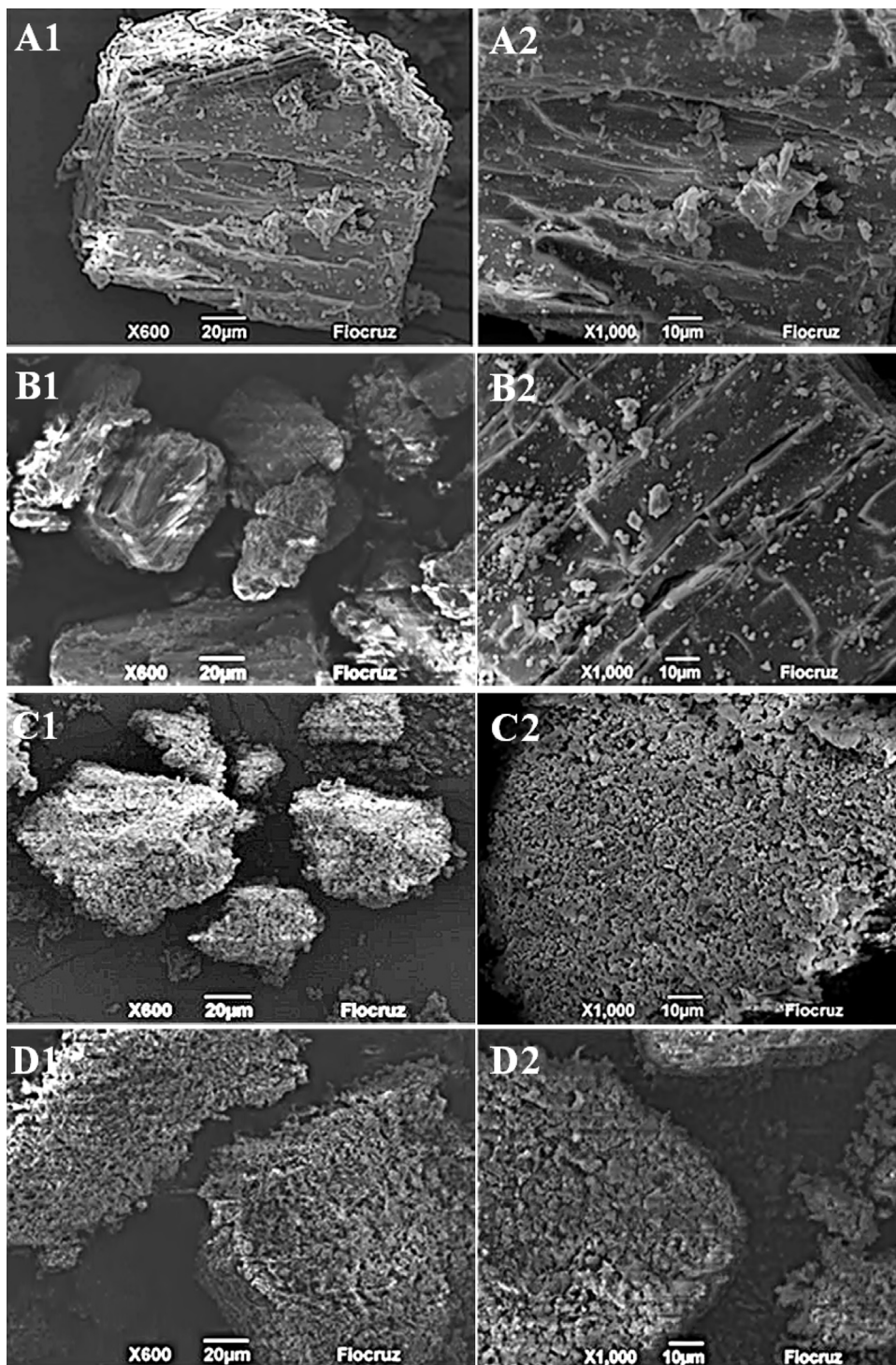
Fig. 2. TG/DTG curves in dynamic  $N_2$  atmosphere of EOHM,  $\beta$ -CD, PM and the complex EOHM: $\beta$ -CD obtained from PC and SC methods.

**Table 1**

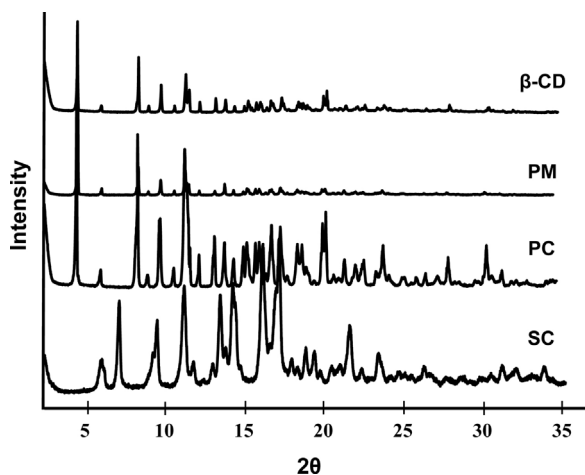
Mass loss percentage obtained by TG of EOHM, PM, PC and SC and Water percentages calculated by Karl Fischer titration.

Sample	% H <sub>2</sub> O (KF)	$\Delta m_1$ (%) 30–170 °C	$\Delta m_2$ (%) 170–280 °C	$\Delta m_3$ (%) 280–400 °C	$\Delta m_4$ (%) 400–900 °C
EOHM	0.52 ± 0.03	100.0	–	–	–
$\beta$ -CD	13.42 ± 0.69	12.2	0.3	65.1	22.4
PM	13.55 ± 0.58	18.4	0.5	69.1	12.0
PC	12.00 ± 0.38	13.1	6.0	84.2	6.7
SC	12.59 ± 0.24	10.2	6.9	70.1	12.8

Fig. 3 shows the SEM images of  $\beta$ -CD, PM, PC and SC, in different magnifications. The  $\beta$ -CD showed different sizes crystals of rectangular shape with some particles adhered on its surface, as already described by other authors [33,34]. In referring to the PM micrograph, no difference was observed between the particles shape compared with the  $\beta$ -CD, suggesting the low complexing ability of this method as reported by the thermoanalytical techniques described above. As for the PC and SC samples, they exhibited different forms when compared to PM and  $\beta$ -CD. These preparations appeared in the form of pellets with a porous surface



**Fig. 3.** Photomicrographs of cross-sections (20 and 10  $\mu$ m, 1 and 2, respectively) of A: $\beta$ -CD, B:PM, C and D: EOHM: $\beta$ -CD obtained by PC and SC methods, respectively.



**Fig. 4.** X-ray diffractograms of  $\beta$ -CD, PM and the complex EOHM: $\beta$ -CD obtained from PC and SC methods.

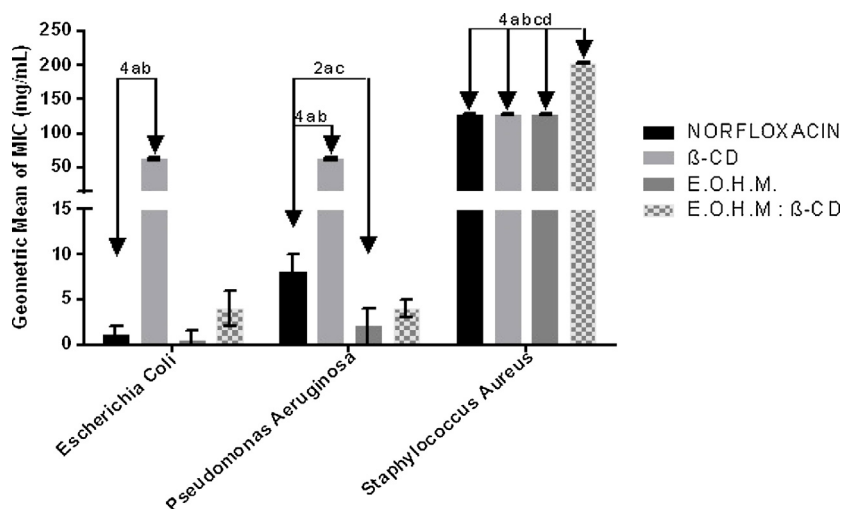
**Table 2**  
Minimum inhibitory concentration values ( $\mu\text{g/ml}$ ).

Substances	Bacteria					
	S.A. ATCC 25923	P.A. ATCC 9027	E.C. ATCC 25922	S.A. 10	P.A. 15	E.C. 06
$\beta$ -CD	$\geq 1024$	$\geq 1024$	$\geq 1024$	$\geq 1024$	$\geq 1024$	$\geq 1024$
EOHM	32	$\geq 1024$	$\geq 1024$	$\geq 1024$	$\geq 1024$	$\geq 1024$
EOHM: $\beta$ -CD	$\geq 1024$	$\geq 1024$	$\geq 1024$	$\geq 1024$	$\geq 1024$	$\geq 1024$

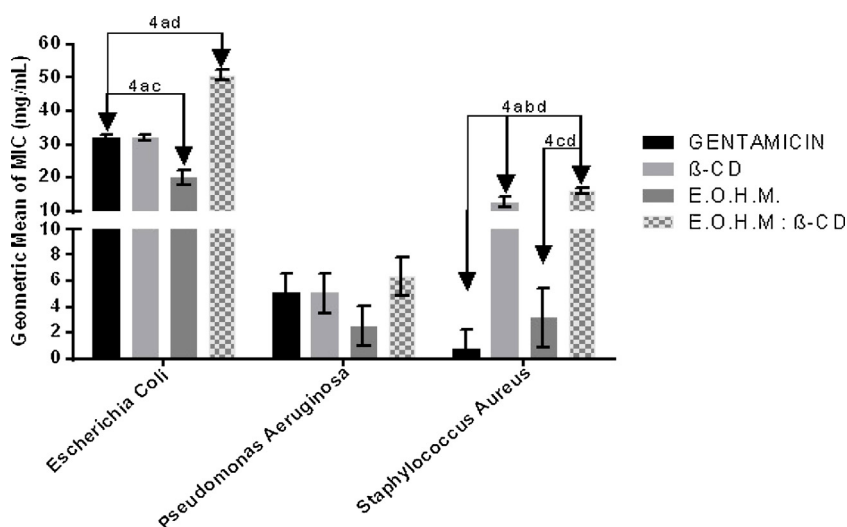
S.A. – *Staphylococcus aureus*; P.A. – *Pseudomonas aeruginosa*; E.C. – *Escherichia coli*;  $\beta$ -CD –  $\beta$ -cyclodextrin; EOHM – essential oil of *Hyptis martiusii*.

revealing an apparent interaction in the solid state between the EOHM and  $\beta$ -CD.

Additionally, X-ray diffraction analysis (Fig. 4) has observed the crystalline nature of the samples. Fig. 4 shows that the  $\beta$ -CD had many reflections, features of its crystalline structure. In the PM was observed some  $\beta$ -CD characteristics with substantial reductions in the intensity, suggesting amorphization process. PC and SC were seen some  $\beta$ -CD reflections and the appearance of new peaks,



**Fig. 5.** Modulatory effect of  $\beta$ -CD, EOHM and the complex EOHM:  $\beta$ -CD in the antibiotic activity of norfloxacin against multidrug resistant strains of *S. aureus* 10, *E. coli* 06 and *P. aeruginosa* 15. Control Norfloxacin (a),  $\beta$ -CD (b), EOHM (c) and EOHM:  $\beta$ -CD (d). The numbers 1–4 express the significance of the association between these substances + norfloxacin. The number 4 means  $P < 0.0001$  when compared with the control.



**Fig. 6.** Modulatory effect of  $\beta$ -CD, EOHM and the complex EOHM:  $\beta$ -CD in the antibiotic activity of gentamicin against multidrug resistant strains of *S. aureus* 10, *E. coli* 06 and *P. aeruginosa* 15. Control gentamicin (a),  $\beta$ -CD (b), EOHM (c) and EOHM:  $\beta$ -CD (d). The numbers 1–4 express the significance of the association between these substances + gentamicin. The number 4 means  $P < 0.0001$  when compared with the control.

which indicated the formation of a new solid phase, thus indicating the formation of the inclusion complex EOHM/ $\beta$ -CD. Similar results were obtained by Toropainen [35].

#### 4.2. Antibacterial activity

The antimicrobial activity was evaluated by Minimum Inhibitory Concentration – MIC (Table 2). The MICs for all compounds were  $\geq 1024 \mu\text{g/mL}$ , except when the essential oil of *Hyptis Martiusii* was used against *Staphylococcus aureus* ATCC 25923 (MIC =  $32 \mu\text{g/mL}$ ), according with previous work by Coutinho et al. [12].

The essential oils of several plants from the genus *Hyptis* demonstrated the best result against Gram-positive bacteria. This fact can be explained due the external membrane present on Gram-negative bacteria, with suppress the association of the lipophilic compounds of the essential oil with the cell membrane [36,37].

Besides all effects reported as solubility enhancement, better adherence to the bacterial cell wall and higher bioavailability of the drug [38], the  $\beta$ -CD alone or associated with the EOHM demonstrated a MIC  $\geq 1024 \mu\text{g/mL}$ . Assays using gamma-CD revealed none effect against none strains of *S. aureus*, *E. coli* and *P. aeruginosa* [39] demonstrating that CD has not an increased antibiotic activity.

#### 4.3. Antibiotic modulatory effect

The EOHM demonstrated a synergism when in association with norfloxacin against *P. aeruginosa* 15, with a MIC reduction of 8 for  $2 \mu\text{g/mL}$  (Fig. 5). This effect can be explained by the high amount of terpenes in the EOHM that enhances the influx of antibiotics by alterations in the cell membrane permeability [40].

$\beta$ -CD demonstrated different antagonistic effect when associated with the antibiotics and against all bacteria assayed. This antagonism effect from the chelating effect of the  $\beta$ -CD with the antibiotics [41,42].

The complex EOHM:  $\beta$ -CD demonstrated higher MICs when compared with the EOHM alone, demonstrating that interaction with the  $\beta$ -CD affected the modulatory effect of the EOHM. Thus, the inclusion complexes demonstrated different physicochemical properties in comparison with the pure compounds, indicating the existence of new manner or molecular organization and chemical interaction with cell systems [43].

Associated with the gentamicin, the EOHM demonstrated synergism against *E. coli* 06, reducing the MIC of  $32\text{--}20.15 \mu\text{g/mL}$  (Fig. 6). Similar results were observed in the essential oil of *Cymbopogon citratus*, a natural complex rich in compounds as terpenes that reduced the MIC of *E. coli* of  $156.25\text{--}39.06 \mu\text{g/mL}$  [44].

Other works using natural products from *Hyptis Martiusii* as the ethanol extract demonstrated a synergism against the strain *E. coli* 27. This fact was associated with the inhibition of efflux systems identified in this bacterium by the use of chlorpromazine (CPZ), causing the same synergistic effect [45] and revealing the possible mechanism of antibiotic activity enhancement.

The  $\beta$ -CD did not demonstrate any modulatory effect against the assayed Gram negative bacteria using gentamicin, however, against the Gram positive one, the effect was an antagonism. When associated with the gentamicin, the complex EOHM:  $\beta$ -CD demonstrated antagonism against the Gram negative and Gram positive bacteria assayed. When compared with the essential oil alone, we can observe that the complexation with  $\beta$ -CD affect directly the modulatory activity of the EOHM [46,47].

## 5. Conclusions

Based on these results, it was possible to observe that the complexation of EOHM in  $\beta$ -CD cavity occurred in the preparations of PC and SC. Additionally, the second stage of TG ( $170\text{--}280^\circ\text{C}$ ) showed that the SC was the best method since the inclusion showed a weight loss of 6.9%, suggesting that the complexed oil required a higher temperature to decompose compared to free EOHM, that lost 100% of its mass up to  $170^\circ\text{C}$ .

The results obtained demonstrated the anti-staphylococcal activity of EOHM and a synergistic effect when associated with the gentamicin against the Gram negative bacteria. Indeed, this essential oil is a potential source of compounds to be used in the treatment of bacterial infections. However, the complex EOHM:  $\beta$ -CD and  $\beta$ -CD alone have not antibacterial neither modulatory antibiotic activity. This result is related with the  $\beta$ -CD do not interact directly with the lipid bilayer as occur with the EOHM alone. Due this fact, the  $\beta$ -CD do not disrupt neither affect the cell membrane.

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